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Periodic Table of the Elements

H₂ P6 ₃ /mmc 3.776 — 6.162																	α-He P6 ₃ /mmc 3.531 — 5.693																												
Li Im3m 3.510 — —	α-Be P6 ₃ /mmc 2.287 — 3.583																	α-B R3m rhomb. 5.057 — —	C Fd3m 3.567 — —	α-N₂ P2 ₁ 3 5.644 — —	α-O₂ C2/m 5.403 — 3.429 — 5.086	α-F₂ C2/m — — —	Ne Fm3m 4.455 — —																						
α-Na P6 ₃ /mmc 3.767 — 6.154	Mg P6 ₃ /mmc 3.209 — 5.210	<div style="display: flex; align-items: center; gap: 10px;"> <div style="border: 1px solid black; padding: 2px;">Element</div> <div style="border: 1px solid black; padding: 2px;">Space group symbol</div> <div style="border: 1px solid black; padding: 2px;">Lattice constant a</div> <div style="border: 1px solid black; padding: 2px;">Lattice constant b</div> <div style="border: 1px solid black; padding: 2px;">Lattice constant c</div> </div>																Al Fm3m 4.050 — —	Si Fd3m 5.431 — —	P Cmca 3.314 — 10.478 — 4.376	α-S₈ Fddd 10.465 — 12.866 — 24.486	Cl₂ Cmca 6.24 — 4.48 — 8.26	Ar Fm3m 5.311 — —																						
K Im3m 5.32 — —	α-Ca Fm3m 5.588 — —	α-Sc P6 ₃ /mmc 3.309 — 5.273	α-Ti P6 ₃ /mmc 2.951 — 4.684	V Im3m 3.024 — —	Cr Im3m 2.885 — —	α-Mn I43m 8.914 — —	α-Fe Im3m 2.866 — —	α-Co Fm3m 3.544 — —	Ni Fm3m 3.524 — —	Cu Fm3m 3.615 — —	Zn P6 ₃ /mmc 2.664 — 7.657 — 4.526	α-Ga Cmca 4.519 — 5.658 — 4.132	Ge Fd3m 5.658 — —	α-As R3m 4.132 — —	Se P3 ₁ 21 4.366 — 4.959	Br₂ Cmca 6.737 — 4.548 — 8.761	Kr Fm3m 5.721 — —																												
Rb Im3m 5.700 — —	α-Sr Fm3m 6.085 — —	α-Y P6 ₃ /mmc 3.647 — 5.731	α-Zr P6 ₃ /mmc 3.232 — 5.148	Nb Im3m 3.299 — —	Mo Im3m 3.147 — —	Tc P6 ₃ /mmc 2.743 — 4.400	Ru P6 ₃ /mmc 2.706 — 4.281	Rh Fm3m 3.804 — —	Pd Fm3m 3.891 — —	Ag Fm3m 4.086 — —	Cd P6 ₃ /mmc 2.979 — 5.619	In I4/mmm 3.253 — 6.489	α-Sn Fd3m 6.489 — —	Sb R3m 4.308 — 11.247	Te P3 ₁ 21 4.457 — 5.927	I₂ Cmca 7.265 — 4.786 — 9.791	Xe Fm3m 6.197 — —																												
Cs Im3m 6.14 — —	Ba Im3m 5.025 — —	α-La P6 ₃ /mmc 3.770 — 12.159	α-Hf P6 ₃ /mmc 3.195 — 5.051	Ta Im3m 3.303 — —	W Im3m 3.165 — —	Re P6 ₃ /mmc 2.761 — 4.458	Os P6 ₃ /mmc 2.735 — 4.319	Ir Fm3m 3.839 — —	Pt Fm3m 3.924 — —	Au Fm3m 4.078 — —	α-Hg R3m 2.993 — —	α-Tl P6 ₃ /mmc 3.456 — 5.525	Pb Fm3m 4.950 — —	Bi R3m 4.546 — 11.862	α-Po Pm3m 3.352 — —	At — — —	Ra — — —																												
Fr	Ra	Ac Fm3m 5.311 — —	<table border="1" style="width: 100%; border-collapse: collapse; text-align: center;"> <tr> <td>α-Ce Fm3m 4.85 — —</td> <td>α-Pr P6₃/mmc 3.673 — 11.835</td> <td>α-Nd P6₃/mmc 3.658 — 11.799</td> <td>Pm — — —</td> <td>α-Sm R3m 8.996 — —</td> <td>Eu Im3m 4.582 — —</td> <td>Gd P6₃/mmc 3.636 — 5.783</td> <td>α-Tb P6₃/mmc 3.601 — 5.694</td> <td>Dy P6₃/mmc 3.590 — 5.648</td> <td>α-Ho P6₃/mmc 3.577 — 5.616</td> <td>α-Er P6₃/mmc 3.559 — 5.587</td> <td>α-Tm P6₃/mmc 3.538 — 5.555</td> <td>α-Yb Fm3m 5.486 — —</td> <td>α-Lu P6₃/mmc 3.503 — 5.551</td> </tr> <tr> <td>α-Th Fm3m 5.084 — —</td> <td>Pa I4/mmm 3.932 — 3.238</td> <td>α-U Cmcm 2.848 — 5.858 — 4.946</td> <td>α-Np Pmcm 4.723 — 4.887 — 6.663</td> <td>α-Pu P2/m 6.183 — 4.822 — 10.963</td> <td>α-Am P6₃/mmc 3.468 — —</td> <td>Cm P6₃/mmc 3.496 — 11.331</td> <td>Bk — — —</td> <td>Cf — — —</td> <td>Es — — —</td> <td>Fm — — —</td> <td>Md — — —</td> <td>No — — —</td> <td>Lw — — —</td> </tr> </table>															α-Ce Fm3m 4.85 — —	α-Pr P6 ₃ /mmc 3.673 — 11.835	α-Nd P6 ₃ /mmc 3.658 — 11.799	Pm — — —	α-Sm R3m 8.996 — —	Eu Im3m 4.582 — —	Gd P6 ₃ /mmc 3.636 — 5.783	α-Tb P6 ₃ /mmc 3.601 — 5.694	Dy P6 ₃ /mmc 3.590 — 5.648	α-Ho P6 ₃ /mmc 3.577 — 5.616	α-Er P6 ₃ /mmc 3.559 — 5.587	α-Tm P6 ₃ /mmc 3.538 — 5.555	α-Yb Fm3m 5.486 — —	α-Lu P6 ₃ /mmc 3.503 — 5.551	α-Th Fm3m 5.084 — —	Pa I4/mmm 3.932 — 3.238	α-U Cmcm 2.848 — 5.858 — 4.946	α-Np Pmcm 4.723 — 4.887 — 6.663	α-Pu P2/m 6.183 — 4.822 — 10.963	α-Am P6 ₃ /mmc 3.468 — —	Cm P6 ₃ /mmc 3.496 — 11.331	Bk — — —	Cf — — —	Es — — —	Fm — — —	Md — — —	No — — —	Lw — — —
α-Ce Fm3m 4.85 — —	α-Pr P6 ₃ /mmc 3.673 — 11.835	α-Nd P6 ₃ /mmc 3.658 — 11.799	Pm — — —	α-Sm R3m 8.996 — —	Eu Im3m 4.582 — —	Gd P6 ₃ /mmc 3.636 — 5.783	α-Tb P6 ₃ /mmc 3.601 — 5.694	Dy P6 ₃ /mmc 3.590 — 5.648	α-Ho P6 ₃ /mmc 3.577 — 5.616	α-Er P6 ₃ /mmc 3.559 — 5.587	α-Tm P6 ₃ /mmc 3.538 — 5.555	α-Yb Fm3m 5.486 — —	α-Lu P6 ₃ /mmc 3.503 — 5.551																																
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Face-centered cubic structure: Fm3m, O_h^h
 Body-centered cubic structure: Im3m, O_h^h
 Hexagonal close packing: P6₃/mmc, D_{6h}^h
 Rhombohedral structure: R3m, D_{3d}^h
 Diamond structure: Fd3m, O_h^h
 Right- and left-handed selenium structure: P3₁21, D₃; P3₂21, D₃

Source: Landolt-Börnstein, New Series Vol. III, b
 Structure Data of Elements and Intermetallic Phases
 (Springer, Berlin, Heidelberg 1971)

Table of Constants

Quantity	Symbol	Value	SI Unit
Velocity of light in vacuum	c	2.997925	10^8 m s^{-1}
Permeability of vacuum	$\mu_0 = 1/\epsilon_0 c^2$	4π	10^{-7} Vs/Am
Permittivity of vacuum	$\epsilon_0 = 1/\mu_0 c^2$	8.854188	10^{-12} As/Vm
Proton rest mass	m_p	1.672649	10^{-27} kg
Electron rest mass	m_e	9.109534	10^{-31} kg
Elementary charge	e	1.602189	10^{-19} As
Planck's constant	h	6.626176	10^{-34} J s
Planck's constant	$\hbar = h/2\pi$	1.054589	10^{-34} J s
Flux quantum	$\Phi_0 = h/2e$	2.067851	$10^{-15} \text{ J A}^{-1}$
Sommerfeld fine-structure constant	$\alpha = \mu_0 c e^2 / 2h$	7.297351	10^{-3}
Rydberg-constant	$R_\infty = \mu_0^2 m_e e^4 c^3 / 8h^3$	1.097373	10^7 m^{-1}
Bohr magneton	$\mu_B = e\hbar/2m_e$	9.274078	10^{-24} Am^2
Avogadro's number	N_A	6.022045	10^{23} mol^{-1}
Atomic mass unit	$m_u = 1 \text{ u} = (10^{-3} \text{ kg mol}^{-1})/N_A$	1.660566	10^{-27} kg
Boltzmann's constant	k	1.380662	$10^{-23} \text{ J K}^{-1}$

Equivalent Values

Quantity	Symbol	Value	SI Unit
Energy equivalent of the Hertz	$E[\text{Hz}] = (1 \text{ Hz})h$	6.626176	10^{-34} J
Frequency equivalent of the electron volt	$\nu[\text{eV}] = 1 \text{ eV}/h$	2.417970	10^{14} Hz
Energy equivalent of the reciprocal meter	$E[\text{m}^{-1}] = (1 \text{ m}^{-1})hc$	1.986478	10^{-25} J
Wave-number equivalent of the electron volt	$\sigma[\text{eV}] = 1 \text{ eV}/hc$	8.065479	10^5 m^{-1}
Energy equivalent of the Kelvin	$E[\text{K}] = (1 \text{ K})k$	1.380662	10^{-23} J
Temperature equivalent of the electron volt	$T[\text{eV}] = 1 \text{ eV}/k$	1.160450	10^4 K